

Question 4

This question relates to Block 2, Chapter 8, and carries 26% of the marks for this assignment.

(a) (11 marks) Consider a carbonaceous chondrite of C2 composition.

(i) If you were presented with a complete, unbroken specimen of such a sample, how would you decide whether it was a meteorite just from its outer appearance? (1 sentence)

(ii) How could you confirm that it was a meteorite by making geochemical measurements? (2 or 3 sentences)

(iii) List the different constituents of a C2 meteorite.

(iv) Why are carbonaceous chondrites in general considered to be 'primitive'? (1 or 2 sentences)

(b) (6 marks) Some carbonaceous chondrites are more primitive than C2 meteorites.

(i) What are they? (1 sentence)

(ii) Explain how these more primitive samples document hydrothermal activity on primitive asteroidal bodies. (2 or 3 sentences)

(iii) What feature of the alteration process has enabled the preservation of the primitive chemical composition? (1 sentence)

(c) (4 marks) Carbonaceous chondrites are so-called because, compared to other meteorites, they contain relatively high proportions of carbon.

(i) What extraterrestrial material, available for study on Earth, is even more carbon-rich than this? (1 sentence)

(ii) What is the likely origin of this carbon-rich material? (1 sentence)

(d) (5 marks) Diamond of interstellar origin is found in C2 meteorites at a concentration of about 0.05% by mass with, on average, only one crystal in 10^{-20} kg of material. Assuming that a C2 meteoroid became fragmented into its constituent grains, say by a collisional event in the asteroid belt, determine whether the diamond grains could eventually spiral in towards the Sun and thus possibly enter the Earth's atmosphere?

(Note: In order to answer this question properly it will be necessary to calculate the size of the diamond grains – use the average size. For this you will need to know the density of diamond, which is 3.5×10^3 kg m⁻³. Show your working.)